

FINAL JEE–MAIN EXAMINATION – APRIL, 2023

(Held On Tuesday 11 April, 2023)

TIME : 3 : 00 PM to 6 : 00 PM

MATHEMATICS

SECTION-A

1. If $\frac{x+1}{x}, \frac{x}{x+1}, \frac{x}{x+2}$ are the roots of the equation $x^3 - 2x^2 + 3x - 4 = 0$, then $|l|$ is

↓ are the roots of the equation

(1) $x^3 - 2x^2 + 3x - 4 = 0$ (2) $x^3 - 2x^2 + 3x + 4 = 0$
(3) $x^3 + 2x^2 + 3x - 4 = 0$ (4) $x^3 - 2x^2 - 3x - 4 = 0$

Official Ans. by NTA (2)

Sol. Put $x = \frac{1}{t}$

$$\frac{1}{t^3} - 2\frac{1}{t^2} + 3\frac{1}{t} - 4 = 0$$

$$1 - 2t + 3t^2 - 4t^3 = 0$$

$$4t^3 - 3t^2 + 2t - 1 = 0$$

Required equation is $4t^3 - 3t^2 + 2t - 1 = 0$

$$x^3 - 2x^2 + 3x - 4 = 0$$

2. Let the line passing through the points, P(2, -1, 2) and Q(0, 3, 4) meet the plane $x - y + z = 4$ at the point R. Then the distance of the point R from the plane $x + 2y + 3z + 2 = 0$ measured parallel to the

line $\frac{x-2}{2} = \frac{y+1}{-3} = \frac{z-2}{2}$ is equal to

(1) $\frac{3}{2}$ (2) $\frac{1}{\sqrt{2}}$
(3) $\frac{1}{2}$ (4) $\frac{3}{\sqrt{2}}$

Official Ans. by NTA (4)

Sol. Line : $\frac{x-2}{2} = \frac{y+1}{-3} = \frac{z-2}{2} = l$

$$R(2+2l, -1-3l, 2+2l)$$

$$3l+5-4l-3+2l+4 = 4$$

$$l+6=4 \Rightarrow l=-2$$

$$R(0, 5, 0)$$

TEST PAPER WITH SOLUTION

Line : $\frac{x+1}{2} = \frac{y+0}{2} = \frac{z-1}{1} = \mu$

Point T = $(2\mu-1, 2\mu, \mu)$

It lies on plane

$$2\mu-1+2(2\mu)+\mu=0$$

$$\mu = -\frac{1}{5}$$

$$T = (-1, -\frac{2}{5}, -\frac{1}{5})$$

$$RT = \frac{1}{\sqrt{5}}$$

3. If the 1011th term from the end in the binomial expansion of $(x^2 - \frac{1}{2x})^{2022}$ is 1024 times 1011th term from the beginning, then $|x|$ is equal to

(1) $\frac{1}{2}$ (2) $\frac{1}{4}$
(3) $\frac{1}{8}$ (4) $\frac{1}{16}$

Official Ans. by NTA (3)

Sol. T₁₀₁₁ from beginning T₁₀₁₁₊₁

$$= {}^{2022}C_{1011} x^{2(2022-1011)} \left(-\frac{1}{2x}\right)^{1011}$$

T₁₀₁₁ from end

$$= {}^{2022}C_{1011} x^{2(1011)} \left(-\frac{1}{2x}\right)^{1011}$$

Given : ${}^{2022}C_{1011} x^{2(2022-1011)} \left(-\frac{1}{2x}\right)^{1011} = 1024 \times {}^{2022}C_{1011} x^{2(1011)} \left(-\frac{1}{2x}\right)^{1011}$

$$= 2^{10} \times x^{2(2022-1011)} \left(-\frac{1}{2x}\right)^{1011} = 2^{10} \times x^{2(1011)} \left(-\frac{1}{2x}\right)^{1011}$$

$$\frac{x^{2(2022-1011)}}{2^{1011}} = 2^{10} \times \frac{x^{2(1011)}}{2^{1011}}$$

$$x^2 = 2^{11}$$

$$x = \sqrt{2^{11}}$$

4. Let the function $f: \mathbb{R} \rightarrow \mathbb{R}$ be defined as

$$f(x) = \begin{cases} \min\{x, [x]\} & \text{if } x \geq 0 \\ e^{-x} - \log_e x & \text{if } x > 0 \end{cases}$$

where $[t]$ denotes the greatest integer less than or equal to t . Then the value of the integral

$$\int_0^2 f(x) dx \text{ is}$$

- (1) $2e - 1$ (2) $1 + \frac{e}{2}$
 (3) $2e - \frac{1}{2}$ (4) $(e - 1) + \frac{1}{2}$

Official Ans. by NTA (3)

Sol. Minimum of $\{x, [x]\} = x$ if $x \in [0, 1)$
 $= x - \log_e x$ if $x \in [1, 2)$

$$f(x) = \begin{cases} x & \text{if } x \in [0, 1) \\ e^{-x} - \log_e x & \text{if } x \in [1, 2) \end{cases}$$

$$\int_0^2 f(x) dx = \int_0^1 x dx + \int_1^2 (e^{-x} - \log_e x) dx$$

$$= \frac{1}{2}(e - 1) + \frac{1}{2}(1 - e)$$

$$= 2e - \frac{1}{2}$$

5. Let $y = y(x)$ be the solution of the differential

$$\text{equations } \frac{dy}{dx} + \frac{y}{x} = \frac{(x^2 + y^2)^{3/2}}{x^2} \quad \text{If } y(1) = 2, \text{ then } y(2) \text{ is equal to}$$

(1) $\frac{13\sqrt{2}}{12\sqrt{3}}$ (2) $\frac{17\sqrt{2}}{12\sqrt{3}}$

- (3) $\frac{19\sqrt{2}}{12\sqrt{3}}$ (4) $\frac{19\sqrt{2}}{12\sqrt{3}}$

Official Ans. by NTA (3)

$$\text{Sol. I.F.} = e^{\int \frac{1}{x} dx} = e^{\log x} = x$$

$$\text{Put } 1 + x^2 = t \Rightarrow 2x dx = dt$$

$$x e^{\int \frac{1}{x} dx} = \frac{1}{t} = \frac{x}{1+x^2}$$

$$y \frac{x}{1+x^2} = \frac{x}{(1+x^2)^{3/2}} \Rightarrow \int \frac{y dx}{1+x^2} = \int \frac{1}{(1+x^2)^{3/2}} dx$$

$$y \times \frac{x}{1+x^2} = \frac{x}{1+x^2} - \frac{1}{1+x^2} + C$$

$$\text{Given that } x=1 \Rightarrow y=2$$

$$2 \times \frac{1}{1+1} = \frac{1}{1+1} - \frac{1}{1+1} + C$$

$$C = \frac{1}{2}$$

$$y \times \frac{x}{1+x^2} = \frac{x}{1+x^2} - \frac{1}{1+x^2} + \frac{1}{2}$$

$$\text{Now put } x=2$$

$$y \times \frac{2}{1+4} = \frac{2}{1+4} - \frac{1}{1+4} + \frac{1}{2}$$

$$y = \frac{19\sqrt{2}}{12\sqrt{3}}$$

6. If four distinct points with position vectors $\vec{a}, \vec{b}, \vec{c}$ and \vec{d} are coplanar, then $[\vec{a}, \vec{b}, \vec{c}]$ is equal to

$$\begin{aligned} (1) & \vec{a} \cdot (\vec{b} \times \vec{c}) + \vec{b} \cdot (\vec{c} \times \vec{d}) + \vec{c} \cdot (\vec{d} \times \vec{a}) + \vec{d} \cdot (\vec{a} \times \vec{b}) \\ (2) & \vec{a} \cdot (\vec{b} \times \vec{c}) + \vec{b} \cdot (\vec{c} \times \vec{d}) + \vec{c} \cdot (\vec{d} \times \vec{a}) + \vec{d} \cdot (\vec{a} \times \vec{b}) \\ (3) & \vec{a} \cdot (\vec{b} \times \vec{c}) + \vec{b} \cdot (\vec{c} \times \vec{d}) + \vec{c} \cdot (\vec{d} \times \vec{a}) + \vec{d} \cdot (\vec{a} \times \vec{b}) \\ (4) & \vec{a} \cdot (\vec{b} \times \vec{c}) + \vec{b} \cdot (\vec{c} \times \vec{d}) + \vec{c} \cdot (\vec{d} \times \vec{a}) + \vec{d} \cdot (\vec{a} \times \vec{b}) \end{aligned}$$

Official Ans. by NTA (1)

Sol. $\vec{a}, \vec{b}, \vec{c}, \vec{d}$ are coplanar points.

$\vec{b} - \vec{a}, \vec{c} - \vec{a}, \vec{d} - \vec{a}$ are coplanar vectors.

$$\text{So, } (\vec{b} - \vec{a}) \cdot (\vec{c} - \vec{a}) \cdot (\vec{d} - \vec{a}) = 0$$

$$(\vec{b} - \vec{a}) \cdot ((\vec{c} - \vec{a}) \times (\vec{d} - \vec{a})) = 0$$

$$\vec{b} \cdot (\vec{c} \times \vec{d}) - \vec{a} \cdot (\vec{c} \times \vec{d}) - \vec{c} \cdot (\vec{a} \times \vec{d}) + \vec{a} \cdot (\vec{b} \times \vec{d}) = 0$$

$$\vec{b} \cdot (\vec{c} \times \vec{d}) - \vec{a} \cdot (\vec{c} \times \vec{d}) - \vec{c} \cdot (\vec{a} \times \vec{d}) + \vec{a} \cdot (\vec{b} \times \vec{d}) = 0$$

7. If $f: \mathbb{R} \rightarrow \mathbb{R}$ be a continuous function satisfying

$$\int_0^{\pi/2} f(\sin x) \sin x dx + \int_0^{\pi/2} f(\cos x) \cos x dx = 0$$

then $\int_0^{\pi/2} f(x) dx$ is equal to

- (1) $-\sqrt{2}$ (2) $\sqrt{2}$
 (3) $\sqrt{2}$ (4) $-\sqrt{2}$

Official Ans. by NTA (3)

$$\text{Sol. I} = \int_0^{\pi/2} f(\sin x) \sin x dx + \int_0^{\pi/2} f(\cos x) \cos x dx$$

$$+ \int_0^{\pi/2} f(\cos x) \cos x dx$$

Apply king in first part and put $x = \pi/2 - t$ in second part.

$$I = \int_0^{\frac{\pi}{2}} (\cos x \sin x) dx + \int_0^{\frac{\pi}{2}} (\cos x \sin x) dx$$

$$I = \int_0^{\frac{\pi}{2}} \cos x \sin x dx + \int_0^{\frac{\pi}{2}} \cos x \sin x dx$$

$$I = \int_0^{\frac{\pi}{2}} \cos x \sin x dx + \int_0^{\frac{\pi}{2}} \cos x \sin x dx$$

$$I = (a + 1) \int_0^{\frac{\pi}{2}} (\cos x \sin x) dx$$

8. If the system of linear equations
 $vx + 11y + az = 13$
 $ox + \xi y + vZ = b$
 $17ox + 19\xi y + 5vZ = 361$ has infinitely many solutions, then equal to
- (1) ξ
 (2) 0
 (3) $a + b + 2$ is
 (4) 6
 (5) 3

Official Ans. by NTA (1)

Sol. $vx + 11y + az = 13$... (i)
 $ox + \xi y + vZ = b$... (ii)
 $17ox + 19\xi y + 5vZ = 361$... (iii)

(i) $\times 10 + (ii) \times 21 - (iii)$
 $Z(10a + 11\xi v - 5v) = 130 + 21b - 361$
 $10a + 11\xi v - 5v = 0$
 $a = -9$
 $130 - 361 + 21b = 0$
 $b = 11$

$a + b + 2 = \xi$

9. The domain of the function

$f(x) = \frac{1}{\sqrt{[x]^2 - 3[x] - 10}}$ is (where $[x]$ denotes the

greatest integer less than or equal to x)

- (1) $[-4, -2) \cup (0, 4)$ (2) $(-4, -3) \cup (3, 4)$
 (3) $[-4, -2) \cup (3, 4)$ (4) $(-4, -3) \cup (0, 4)$

Official Ans. by NTA (3)

Sol. $[x]^2 - 3[x] - 10 > 0$
 $[x] < -2$ or $[x] > 4$

10. Let P be the plane passing through the points $(0, 3, 0)$, $(1, 3, -2)$ and $(1, 6, 2)$. For $a \in \mathbb{N}$, if the distances of the points A $(3, \xi, a)$ and B (γ, a, a) from the plane P are γ and ξ respectively, then the positive value of a is

- (1) 6 (2) ξ
 (3) 3 (4) 0

Official Ans. by NTA (2)

Sol. $\begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 1 & 0 & -2 \\ 0 & 3 & 2 \end{vmatrix} = \hat{i}(6) + \hat{j}(-2) + \hat{k}(3)$

Normal of the plane $6\hat{i} - 2\hat{j} + 3\hat{k}$

Plane : $3x - \xi y + 1.5z = 3$

Distance from A $(3, \xi, a)$

$\frac{|9 - 11\xi + 1.5a - 3|}{1.5} = \gamma$
 $a = 3$

$a = -1$ (rejected)

Distance from B (γ, γ, a)

$\frac{|6 - 11\xi + 1.5a - 3|}{1.5} = \xi$
 $a = \xi$

11. The converse of the statement $(p \rightarrow q) \rightarrow r$ is
 (1) $(\neg r) \rightarrow p \rightarrow q$ (2) $(\neg r) \rightarrow (p \rightarrow q)$
 (3) $(\neg r) \rightarrow (p \rightarrow q) \rightarrow r$ (4) $(p \rightarrow q) \rightarrow (\neg r)$

Official Ans. by NTA (4)

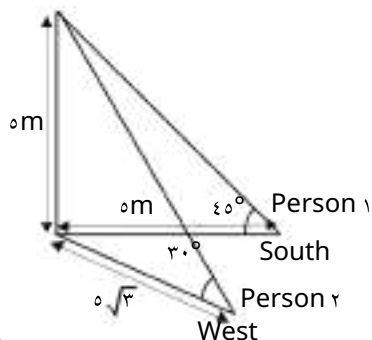
Sol. Converse of $(p \rightarrow q) \rightarrow r$

$\neg r \rightarrow (p \rightarrow q)$
 $\neg r \rightarrow (\neg p \rightarrow q)$
 $\neg r \rightarrow (\neg p \rightarrow q) \rightarrow r$

12. The angle of elevation of the top P of a tower from the feet of one person standing due South of the tower is 30° and from the feet of another person standing due west of the tower is 45° . If the height of the tower is 10 meters, then the distance (in meters) between the two persons is equal to

- (1) 10 (2) 0
 (3) 0 (4) $10\sqrt{5}$

Official Ans. by NTA (1)



Sol. Distance = 10 (By Pythagoras theorem)

13. Let a, b, c and d be positive real numbers such that $a + b + c + d = 11$. If the maximum value of $a^2 b^2 c^2 d^2$ is $3750 \cdot b$, then the value of b is

- (1) 90 (2) 110
(3) 50 (4) 108

Official Ans. by NTA (1)

$$\text{Sol. } \frac{5a^2 b^2 c^2 d^2}{50} + \frac{a^2 b^2 c^2 d^2}{3} + \frac{a^2 b^2 c^2 d^2}{2} + d = \frac{a^2 b^2 c^2 d^2}{3750 \cdot b}$$

$$\frac{a^2 b^2 c^2 d^2}{3750 \cdot b} = 90$$

14. If the radius of the largest circle with centre $(2, 0)$ inscribed in the ellipse $x^2 + 4y^2 = 36$ is r , then $12r$ is equal to

- (1) 72 (2) 110
(3) 92 (4) 69

Official Ans. by NTA (3)

$$\text{Sol. } (x-2)^2 + 4y^2 = r^2$$

Solving with ellipse, we get

$$(x-2)^2 + \frac{36-x^2}{4} = r^2$$

$$3x^2 - 16x + 52 - 4r^2 = 0$$

$$D = 0 \Rightarrow 16^2 - 4 \cdot 3 \cdot (52 - 4r^2) = 0$$

15. Let the mean of 6 observation $1, 2, 4, 5, x$ and y be 6 and their variance be 10. Then their mean deviation about the mean is equal to

- (1) $\frac{10}{3}$ (2) $\frac{7}{3}$
(3) 3 (4) $\frac{8}{3}$

Official Ans. by NTA (4)

$$\text{Sol. } x + y = 18 \quad \text{Q mean} = 6 \dots (i)$$

$$10 = \frac{1+2+4+5+x+y}{6} - 6$$

$$x + y = 16 \dots (ii)$$

By solving (i) and (ii)

$$x = 8, y = 8$$

$$\text{M.D.} = \frac{\sum |x_i - \bar{x}|}{n} = \frac{8}{3}$$

16. The sum of the coefficients of three consecutive terms in the binomial expansion of $(1+x)^n$ which are in the ratio $1:3:5$ is equal to

- (1) 20 (2) 63
(3) 41 (4) 92

Official Ans. by NTA (2)

$$\text{Sol. } {}^{n+r}C_r : {}^{n+r}C_{r+1} : {}^{n+r}C_{r+2} = 1:3:5$$

$$\frac{{}^{n+r}C_{r-1}}{{}^{n+r}C_r} = \frac{r}{r+1}$$

$$n+r-3 \dots (i)$$

$$\frac{{}^{n+r}C_r}{{}^{n+r}C_{r+1}} = \frac{r}{r+1}$$

$$nr-1=3n \dots (ii)$$

From (i) and (ii)

$$r=2 \text{ and } n=5$$

$$\text{Required sum} = 63$$

17. If the letters of the word MATHS are permuted and all possible words so formed are arranged as in a dictionary with serial numbers, then the serial number of the word THAMS is

- (1) 103 (2) 104
(3) 101 (4) 102

Official Ans. by NTA (1)

$$\text{Sol. } 4! + 3! + 2! = 103$$

18. For $a \in \mathbb{C}$, let $A = \{z \in \mathbb{C} : \operatorname{Re}(a+z) < \operatorname{Im}(a+z)\}$ and $B = \{z \in \mathbb{C} : \operatorname{Re}(a+z) > \operatorname{Im}(a+z)\}$. Then

among the two statements :

(S₁) : If $\operatorname{Re}(A) \cap \operatorname{Im}(A) \neq \emptyset$, then the set A contains all the real numbers

(S₂) : If $\operatorname{Re}(A) \cap \operatorname{Im}(A) = \emptyset$, then the set B contains all the real numbers.

- (1) Only (S₁) is true (2) both are false
(3) Only (S₂) is true (4) Both are true

Official Ans. by NTA (2)

Sol. Let $a = x_1 + iy_1, z = x_2 + iy_2$

Now $\text{Re}(a+z) < \text{Im}(a+z)$

$$x_1 + x_2 < -y_1 + y_2$$

$$x_1 = 2, y_1 = 1, x_2 = -1, y_2 = 0$$

Given inequality is not valid for these values.

S_1 is false.

Now $\text{Re}(a+z) > \text{Im}(a+z)$

$$x_1 + x_2 > -y_1 + y_2$$

$$x_1 = -2, y_1 = -1, x_2 = 1, y_2 = 0$$

Given inequality is not valid for these values.

S_2 is false.

19. Let $A = \{1, 2, 3, 4, 5, 6, 7, 8, 9\}$ and $B = \{2, 3, 4, 5, 6, 7, 8, 9, 10\}$. Let R be a relation defined on $A \times B$ such that $R = \{(a_1, b_1), (a_2, b_2), \dots : a_1 \in B_1 \text{ and } b_1 \in A_1\}$. Then the number of elements in the set R is

$$(1) 26$$

$$(2) 16$$

$$(3) 18$$

$$(4) 52$$

Official Ans. by NTA (2)

Sol. Let $a_1 = 1$ choices of b_1

$a_2 = 2$ choices of b_2

$a_3 = 3$ choices of b_3

$a_4 = 4$ choices of b_4

$a_5 = 5$ choices of b_5

For (a_1, b_1) 16 ways.

Similarly, $b_1 = 2$ choices of a_1

$b_2 = 3$ choices of a_2

$b_3 = 4$ choices of a_3

$b_4 = 5$ choices of a_4

Required elements in $R = 16$.

20. Let f and g be two functions defined by

$$f(x) = \begin{cases} x+1, & x > 0 \\ x^3, & x \leq 0 \end{cases} \text{ and } g(x) = \begin{cases} x+1, & x > 0 \\ x^3, & x \leq 0 \end{cases}$$

Then $(g \circ f)(x)$ is

(1) Differentiable everywhere

(2) Continuous everywhere but not differentiable exactly at one point

(3) Not continuous at $x = -1$

(4) Continuous everywhere but not differentiable at $x = 1$

Official Ans. by NTA (2)

$$\text{Sol. } f(x) = \begin{cases} x+1, & x > 0 \\ x^3, & x \leq 0 \end{cases}$$

$$g(x) = \begin{cases} x+1, & x > 0 \\ x^3, & x \leq 0 \end{cases}$$

$$g(f(x)) = \begin{cases} x+2, & x > -1 \\ x^3-1, & x \leq -1 \end{cases}$$

$g(f(x))$ is continuous everywhere

$g(f(x))$ is not differentiable at $x = -1$

Differentiable everywhere else

SECTION-B

The number of points, where the curve

$$f(x) = e^x - e^{-x} - xe^x - e^{-x} + 1, x \in \mathbb{R}$$

is equal to

Official Ans. by NTA (2)

Sol. Let $e^x = t$

$$t^4 - t^3 - t^2 - t + 1 = 0$$

$$t^2 + t^2 - t^3 - t^2 - t + 1 = 0$$

$$t^2 + t^2 - t^3 - t^2 - t + 1 = 0$$

$$t + \frac{1}{t} = \frac{1 + \sqrt{5}}{2}$$

Two real values of t .

22. Let the probability of getting head for a biased coin

be $\frac{1}{\xi}$. It is tossed repeatedly until a head appears.

Let N be the number of tosses required. If the

probability that the equation $16x^2 + 5Nx + 1 = 0$

has no real root is $\frac{p}{q}$, where p and q are co-prime.

then $q - p$ is equal to

Official Ans. by NTA (27)

Sol. $64x^2 + 0Nx + 1 = 0$

$D = 20N^2 - 206 > 0$

$\therefore N^2 < \frac{206}{20} \Rightarrow N < \frac{16}{5}$

$\therefore N = 1, 2, 3$

$\therefore \text{Probability} = \frac{1}{5} + \frac{3}{5} + \frac{1}{5} + \frac{3}{5} + \frac{3}{5} + \frac{1}{5} = \frac{16}{5}$

$\therefore q - p = 27$

23. Let $a = i\hat{a} + j\hat{b} + k\hat{c}$ and $b = i\hat{a} + j\hat{b} - k\hat{c}$. If \hat{a} is a vector such that $\hat{a} \times \hat{c} = \hat{b}$, $\hat{a} \times \hat{b} = \hat{c}$ and $\hat{b} \times \hat{a} = \hat{c}$, then \hat{a} is equal to

Official Ans. by NTA (2018)

Sol. $\hat{a} = i\hat{a} + j\hat{b} + k\hat{c}$ $\hat{b} = i\hat{a} + j\hat{b} - k\hat{c}$

$\hat{a} \times \hat{b} = \hat{a} \times (i\hat{a} + j\hat{b} - k\hat{c}) = \hat{a} \times \hat{b}$

$\hat{b} \times \hat{a} = \hat{c}$

Let θ be angle between \hat{a} and \hat{b}

Then $|\hat{a} \times \hat{b}| = |\hat{a}| |\hat{b}| \sin \theta = \sqrt{13}$

$|\hat{b} \times \hat{a}| = |\hat{b}| |\hat{a}| \cos \theta = 27$

$\therefore \sin \theta = \frac{\sqrt{13}}{27}$

$|\hat{b}| |\hat{a}| \cos \theta = 27 \sqrt{13}$

$\therefore |\hat{a}| |\hat{b}| = \sqrt{13} \sqrt{90}$

24. Let $S = \sum_{n=0}^{\infty} (z^n - i^n) \left(\frac{z^2 + \lambda z - 1}{z - \bar{\lambda} z - \bar{\lambda}} \right)^n$. If

$a - \frac{1}{\lambda} i S, a \in \mathbb{C} \setminus \{0\}$, then $\text{Re}(a)$ is equal to

Official Ans. by NTA (2018)

Sol. $\frac{z^2 + \lambda z - 1}{z - \bar{\lambda} z - \bar{\lambda}} = \frac{z^2 + \lambda z - 1}{z(1 - \bar{\lambda}) - \bar{\lambda}}$

$\therefore 1 + \frac{(1 - \bar{\lambda} z - 1)}{(z - \bar{\lambda} z - \bar{\lambda})} = \frac{z - \bar{\lambda} z - \bar{\lambda}}{z - \bar{\lambda} z - \bar{\lambda}}$

Put $z = a - \frac{1}{\lambda} i$

$\therefore (z - \bar{\lambda} z - \bar{\lambda})$ is imaginary

Put $z = x + iy$

$\therefore (x^2 - y^2 + 2xyi - \bar{\lambda}x + \bar{\lambda}y - \bar{\lambda})$ is Imaginary

$\therefore \text{Re}(x^2 - y^2 + 2xyi - \bar{\lambda}x + \bar{\lambda}y - \bar{\lambda}) = 0$

$\therefore x^2 - y^2 + 2xy - \bar{\lambda}x + \bar{\lambda}y - \bar{\lambda} = 0$

$x^2 = y^2 - 2xy + \bar{\lambda}x - \bar{\lambda}y + \bar{\lambda}$

$x = (y - 1)(y - \bar{\lambda}) \Rightarrow z = a - \frac{1}{\lambda} i$

Put $x = a, y = -\frac{1}{\lambda}$

$a^2 = \frac{1}{\lambda^2} - \frac{1}{\lambda} \frac{1}{\lambda} - \frac{1}{\lambda} \frac{1}{\lambda} = \frac{1}{\lambda^2} - \frac{2}{\lambda^2}$

$a^2 = \frac{1}{\lambda^2} - \frac{2}{\lambda^2} = \frac{1}{\lambda^2}$

$\therefore \text{Re}(a^2) = \frac{1}{\lambda^2} - \frac{2}{\lambda^2} = -\frac{1}{\lambda^2}$

25. For $k \in \mathbb{R}$, if the sum of the series

$1 + \frac{k}{2} + \frac{k^2}{4} + \frac{k^3}{8} + \frac{k^4}{16} + \dots$ is 1, then the value of k

is

Official Ans. by NTA (2018)

Sol. $10 = 1 + \frac{1}{k} + \frac{1}{k^2} + \frac{1}{k^3} + \frac{1}{k^4} + \dots$ upto

$9 = \frac{1}{k} + \frac{1}{k^2} + \frac{1}{k^3} + \frac{1}{k^4} + \dots$ upto

$\frac{9}{k} = \frac{1}{k^2} + \frac{1}{k^3} + \frac{1}{k^4} + \dots$ upto

$S = \frac{1}{k} - \frac{1}{k^2} + \frac{1}{k^3} - \frac{1}{k^4} + \frac{1}{k^5} - \frac{1}{k^6} + \dots$ upto

$S = \frac{1}{k^2} + \frac{1}{k^3} + \frac{1}{k^4} + \dots$ upto

$\frac{1}{k} - \frac{1}{k^2} = \frac{1}{k} + \frac{1}{k^3} + \frac{1}{k^4} + \frac{1}{k^5} + \dots$

$\frac{1}{k} - \frac{1}{k^2} = \frac{1}{k} + \frac{1}{k^3} + \frac{1}{k^4} + \frac{1}{k^5} + \dots$

$9(k-1) = k(k-1) + 1$

$k = 2$

26. Let $A = \{1, 2, 3, 4, 5\}$ and $B = \{1, 2, 3, 4, 5, 6\}$. Then the number of functions $f: A \rightarrow B$ satisfying $f(1) + f(2) = f(3) + 1$ is equal to

Official Ans. by NTA (360)

Sol. $f(1) + f(2) + 1 = f(3) + 1$

$f(1) + f(2) = f(3)$

Case (i) $f(1) = 1$ $f(2) = 1, 2, 3, 4$ \Rightarrow 4 mappings

Case (ii) $f(1) = 2$ $f(2) = 1, 2, 3$ \Rightarrow 3 mappings

Case (iii) $f(1) = 3$ $f(2) = 1, 2$ \Rightarrow 2 mappings

Case (iv) $f(1) = 4$ $f(2) = 1$ \Rightarrow 1 mapping

$f(5)$ & $f(6)$ both have 6 mappings each

Number of functions $= (4 + 3 + 2 + 1) \times 6 \times 6 = 360$ Official Ans. by NTA (11)

27. Let the tangent to the parabola $y^2 = 4x$ at the point (x_1, a) be perpendicular to the line $2x + 3y = 3$. Then the square of distance of the point $(1, -2)$ from the normal to the hyperbola $ax^2 - by^2 = 9a^2$ at its point $(a - 1, a + 2)$ is equal to

Official Ans. by NTA (116)

Sol. $P(x_1, a)$ lies on $y^2 = 4x$

$4a^2 = 4x_1$

But, $\frac{dy}{dx} \bigg|_{(x_1, a)} = \frac{2}{a} = 1 \Rightarrow a = 2$ (a = 1 reject)

Now, hyperbola $\frac{x^2}{9} - \frac{y^2}{36} = 1$, normal at

$Q(a-1, a+2)$ is $\frac{ax}{9} + \frac{3by}{18} = 0$

$2x + 3y = 0$

Now, distance of $(1, -2)$ from $2x + 3y = 0$ is equal to

$\frac{|2(1) - 3(2)|}{\sqrt{2^2 + 3^2}} = \frac{4}{\sqrt{13}}$

\Rightarrow Square of distance is $\frac{16}{13}$

28. Let the line $l: x = \frac{1-y}{-2} = \frac{z-3}{1}$ meet the plane $P: x + 2y + 3z = 4$ at the point (a, b, c) . The angle between the line l and the plane P is

$\cos^{-1} \frac{a}{\sqrt{a^2 + b^2 + c^2}}$ then $a + 2b + 6c$ is equal to

Official Ans. by NTA (11)

Sol. $\vec{l} : x = \frac{y-1}{r} = \frac{z-r}{l} \cdot \hat{l}$

DR's of line $l(x, y, z)$

DR's of normal vector of plane $P : x + 2y + 3z = 1$ are $(1, 2, 3)$

Now, angle between line l and plane P is given by

$$\sin \theta = \frac{|1 + 4 + 9|}{\sqrt{1+4+9} \times \sqrt{1+4+9}} = \frac{14}{14} \Rightarrow \cos \theta = \frac{0}{14} = 0$$

$$\theta = \frac{\pi}{2}$$

Let variable point on line l is $\frac{x}{r}t + \frac{y}{l}t + \frac{z}{l}t + 3 = 0$

lies on plane P .

$$\frac{x}{r}t + \frac{y}{l}t + \frac{z}{l}t + 3 = 0$$

$$\frac{x}{r}t + \frac{y}{l}t + \frac{z}{l}t + 3 = 0$$

$$\frac{x}{r}t + \frac{y}{l}t + \frac{z}{l}t + 3 = 0$$

$$x + 2y + 6z = -11$$

29. If the line $l_1 : 3y - 2x = 3$ is the angular bisector of

the lines $l_2 : x - y + 1 = 0$ and $l_3 : ax + by + 17 = 0$,

then $a^2 + b^2 - a$ is equal to

Official Ans. by NTA (348)

Sol. Point of intersection of $l_1 : 3y - 2x = 3$

$l_2 : x - y + 1 = 0$ is $P(0, 1)$

Which lies on $l_3 : ax + by + 17 = 0$,

$$b = -17$$

Consider a random point $Q(1, 0)$,

on $l_2 : x - y + 1 = 0$, image of Q about

$l_1 : 3y - 2x = 3$ is $Q'(0, 2)$ which is

calculated by formulae

$$\frac{x - (-1)}{2} = \frac{y - 0}{-3} = \frac{z - r}{0} = -2$$

Now, Q' lies on $l_3 : ax + by + 17 = 0$,

$$b = 7$$

Now, $a + b - a - b = 348$

30.

If A is the area in the first quadrant enclosed by

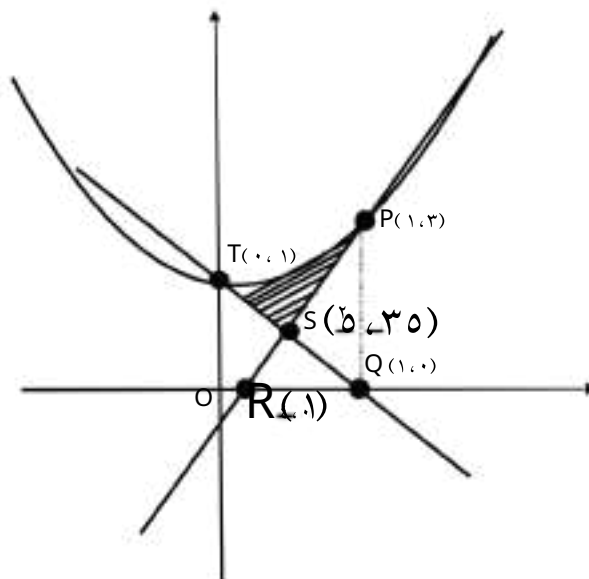
the curve $C : 2x^2 - y + 1 = 0$, the tangent to C at the

point $(1, 3)$ and the line $x + y = 1$, then the value of

$10A$ is

Official Ans. by NTA (16)

Sol.



$$y = 2x^2 + 1$$

Tangent at $(1, 3)$

$$y = 4x - 1$$

$$A = \int_0^1 (2x^2 + 1) dx - \text{area of } \triangle DQOT - \text{area of } \triangle DPQR + \text{area of } \triangle DQRS$$

$$A = \frac{2}{3} + 1 - \frac{1}{2} - \frac{1}{8} + \frac{1}{24} = \frac{17}{24}$$

PHYSICS

SECTION-A

31. Eight equal drops of water are falling through air with a steady speed of 10 cm/s . If the drops coalesce, the new velocity is :-

- (1) 10 cm/s (2) 40 cm/s
(3) 16 cm/s (4) 80 cm/s

Official Ans. by NTA (2)

Sol. $v \propto r^2$

$$\frac{v_1}{v_2} = \left(\frac{r_1}{r_2}\right)^2$$

$$\frac{10}{v_2} = \left(\frac{R}{8R}\right)^2$$

$$R = 8r$$

$$\frac{10}{v_2} = \left(\frac{1}{8}\right)^2$$

$$v_2 = 640 \text{ cm/s}$$

32. A car P travelling at 20 m/s sounds its horn at a frequency of 440 Hz . Another car Q is travelling behind the first car in the same direction with a velocity 40 m/s . The frequency heard by the passenger of the car Q is approximately. Take, velocity of sound = 360 m/s

- (1) 514 Hz (2) 421 Hz
(3) 480 Hz (4) 471 Hz

Official Ans. by NTA (2)

Sol. $f = f_0 \frac{c + v_o}{c - v_s}$

$$f = 440 \times \frac{360 + 40}{360 - 20}$$

$$f = 471 \text{ Hz}$$

TEST PAPER WITH SOLUTION

33. A plane electromagnetic wave of frequency $2 \times 10^{10} \text{ MHz}$ propagates in free space along x-direction. At a particular space and time, $\vec{E} = 6\hat{i} + 3\hat{j} \text{ V/m}$. What is \vec{B} at this point?

- (1) $2 \times 10^{-10} \hat{k} \text{ T}$ (2) $2 \times 10^{-10} \hat{i} \text{ T}$
(3) $2 \times 10^{-10} \hat{j} \text{ T}$ (4) $2 \times 10^{-10} \hat{k} \text{ T}$

Official Ans. by NTA (2)

Sol. $E = 6\hat{i} + 3\hat{j}$
 $2 \times 10^{10} \text{ MHz}$
 $c = 3 \times 10^8 \text{ m/s}$
 $\rightarrow \hat{E} \perp \hat{B} \perp \hat{c}$

$$|\vec{B}| = \frac{|\vec{E}|}{c} = 2 \times 10^{-10} \text{ T}$$

$$\vec{E} \cdot \vec{B} = 0$$

$$\vec{B} = 2 \times 10^{-10} \hat{j} \text{ T}$$

34. A capacitor of capacitance C is charged to a potential V . The flux of the electric field through a closed surface enclosing the positive plate of the capacitor is:

- (1) $\frac{CV}{\epsilon_0}$ (2) $\frac{qV}{\epsilon_0}$
(3) $\frac{CV}{\epsilon_0}$ (4) Zero

Official Ans. by NTA (3)

Sol. $\oint \vec{E} \cdot d\vec{A} = \frac{q_{in}}{\epsilon_0}$

$$\oint \vec{E} \cdot d\vec{A} = \frac{Q}{\epsilon_0}$$

$$\oint \vec{E} \cdot d\vec{A} = \frac{CV}{\epsilon_0}$$

30. If force (F), velocity (V) and time (T) are considered as fundamental physical quantity, then dimensional formula of density will be:

- (1) FVT^{-2} (2) FVT
 (3) FVT^{-1} (4) FVT^{-2}

Official Ans. by NTA (2)

Sol. $[ML^{-3}] = [MLT^{-2}]^a [LT^{-1}]^b [T]^c$

$$= [M]^{a+b} [L]^{-3a-b+c} [T]^{-2a-b+c}$$

$$a = 1$$

$$a + b = -3$$

$$b = -4$$

$$\text{also } -2a - b + c = 0$$

$$c = -2$$

36. In satellite communication, the uplink frequency band used is:

- (1) 3.7 – 4.2 GHz
 (2) 0.925 – 1.425 GHz
 (3) 76 – 88 MHz
 (4) 420 – 490 MHz

Official Ans. by NTA (2)

Sol. Conceptual

37. If V is the gravitational potential due to sphere of uniform density on its surface, then its value at the center of sphere will be:-

- (1) $\frac{rV}{2}$ (2) V
 (3) $\frac{3}{2}V$ (4) $\frac{V}{2}$

Official Ans. by NTA (1)

Sol. $V = \frac{GM}{r}$ at $r = R$, $V = \frac{GM}{R}$

$$\text{at } r = 0, V = \frac{rGM}{2R} = \frac{rV}{2}$$

38. A body of mass 0.1 g moves along x-axis such that its velocity varies with displacement x according to the relation $v = \sqrt{x}$ m/s the force acting on the body is:-

- (1) 166 N (2) 20 N
 (3) 120 N (4) 0 N

Official Ans. by NTA (2)

Sol. $v = \sqrt{x}$

$$\frac{dv}{dx} = \frac{1}{2\sqrt{x}} = \frac{1}{2\sqrt{0.01}} = 5 \text{ m/s}^2$$

$$F = 20 \text{ N}$$

39. A projectile is projected at 30° from horizontal with initial velocity 40 ms^{-1} . The velocity of the projectile at $t = 2 \text{ s}$ from the start will be:

(Given $g = 10 \text{ m/s}^2$)

- (1) 40 ms^{-1} (2) 40 ms^{-1}
 (3) 20 ms^{-1} (4) Zero

Official Ans. by NTA (1)

Sol. At $t = 2$ particle is at maximum height

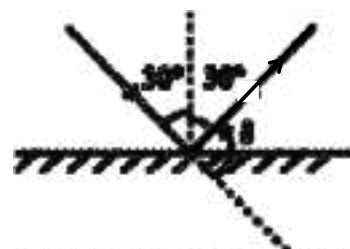
moving with velocity $V = 40 \cos 30^\circ = 20 \text{ ms}^{-1}$

When one light ray is reflected from a plane mirror with 30° angle of reflection, the angle of deviation of the ray after reflection is:

- (1) 120° (2) 120°
 (3) 110° (4) 130°

Official Ans. by NTA (2)

Sol.



$$180^\circ - 2 \times 30^\circ = 120^\circ$$

41. A spaceship of mass 2×10^4 kg is launched into a circular orbit close to the earth surface. The additional velocity to be imparted to the spaceship in the orbit to overcome the gravitational pull will be (radius of earth = 6400 km)

(1) $11.2 \sqrt{2}$ km/s

(2) $11.2 \sqrt{2}$ km/s

(3) $11.2 \sqrt{2}$ km/s

(4) $11.2 \sqrt{2}$ km/s

Official Ans. by NTA (3)

Sol. $v_{\text{orbit}} = \sqrt{\frac{GM}{R}}$

$v_{\text{escape}} = \sqrt{\frac{2GM}{R}} = \sqrt{2} v_{\text{orbit}}$

$v = \sqrt{2} v_{\text{orbit}} = \sqrt{2} \times 11.2 \text{ km/s}$

42. The ratio of the de-Broglie wavelengths of proton and electron having same kinetic energy: (Assume $m_p = m_e \times 1836$)

(1) $1 : 43$

(2) $1 : 43$

(3) $1 : 43$

(4) $1 : 43$

Official Ans. by NTA (1)

Sol. $\frac{1}{\sqrt{m}} \propto \frac{1}{\sqrt{p}} \propto \frac{1}{\sqrt{m_e}}$

43. The thermodynamic process, in which internal energy of the system remains constant is

(1) Isochoric

(2) Isothermal

(3) Adiabatic

(4) Isobaric

Official Ans. by NTA (2)

Sol. $T = \text{constant} \Rightarrow U = \text{constant}$

44. The energy of He ion in its first excited state is. (The ground state energy for the Hydrogen atom is -13.6 eV):

(1) -3.4 eV

(2) -0.85 eV

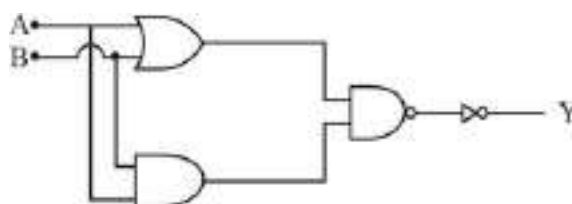
(3) -13.6 eV

(4) -27.2 eV

Official Ans. by NTA (3)

Sol. $E_n = \frac{-13.6 Z^2}{n^2} = \frac{-13.6 \times 4}{4} = -13.6 \text{ eV}$

45. The logic operations performed by the given digital circuit is equivalent to:



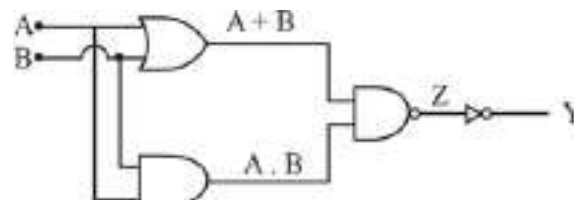
(1) AND

(2) NOR

(3) OR

(4) NAND

Official Ans. by NTA (1)



Sol.

$Z = A + B$

$Y = Z \cdot A \cdot B = (A + B) \cdot A \cdot B$

$Y = A \cdot B$

It is an AND Gate

46. The root mean square speed of molecules of nitrogen gas at 27°C is approximately: (Given mass of a nitrogen molecule = 4.6×10^{-26} kg)

and take Boltzmann constant $k_B = 1.38 \times 10^{-23} \text{ J/K}$

(1) 523 m/s

(2) 1260 m/s

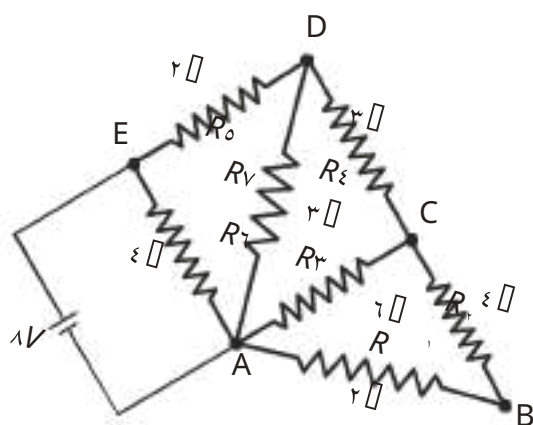
(3) 91 m/s

(4) 27.4 m/s

Official Ans. by NTA (1)

Sol. $V_{\text{rms}} = \sqrt{\frac{3kT}{m}} = \sqrt{\frac{3 \times 1.38 \times 10^{-23} \times 300}{4.6 \times 10^{-26}}} = 523 \text{ m/s}$

87.

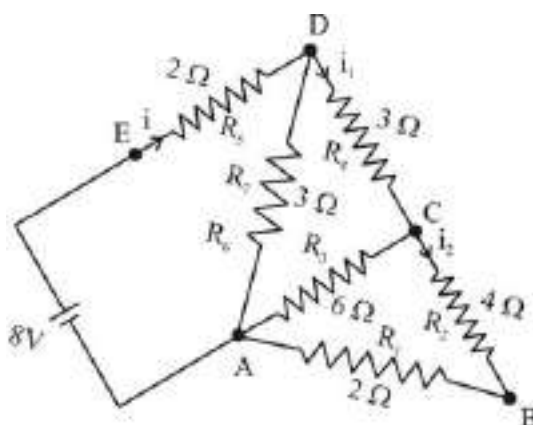


The current flowing through R_1 is:

- (1) $\frac{1}{2}A$
- (2) $\frac{1}{4}A$
- (3) $\frac{1}{8}A$
- (4) $\frac{1}{16}A$

Official Ans. by NTA (4)

Sol.



$$R_{eq} = 4\Omega$$

$$i = \frac{8}{4} = 2A$$

$$i_1 = \frac{3}{3+4} \times 2 = \frac{6}{7}A$$

$$i_2 = \frac{3}{7}A$$

88.

When vector $A = \hat{i} + \hat{j} + \hat{k}$ is subtracted from vector B , it gives a vector equal to \hat{j} . Then the magnitude of vector B will be:

- (1) $\sqrt{13}$
- (2) 3
- (3) $\sqrt{6}$
- (4) $\sqrt{5}$

Official Ans. by NTA (BONUS)

Sol. $B - A = \hat{j}$

$$B = \hat{i} + \hat{j} + \hat{k}$$

$$|B| = \sqrt{3}$$

89.

Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R.
Assertion A: A bar magnet dropped through a metallic cylindrical pipe takes more time to come down compared to a non-magnetic bar with same geometry and mass.

Reason R: For the magnetic bar, Eddy currents are produced in the metallic pipe which oppose the motion of the magnetic bar.

In the light of the above statements, choose the correct answer from the options given below

- (1) Both A and R are true but R is NOT the correct explanation of A
- (2) A is true but R is false
- (3) Both A and R are true and R is the correct explanation of A
- (4) A is false but R is true

Official Ans. by NTA (3)

Sol. Conceptual

50. An electron is allowed to move with constant velocity along the axis of current carrying straight solenoid. A. The electron will experience magnetic force along the axis of the solenoid. B. The electron will not experience magnetic force. C. The electron will continue to move along the axis of the solenoid.

D. The electron will be accelerated along the axis of the solenoid.

E. The electron will follow parabolic path inside the solenoid.

Choose the correct answer from the options given below.

Official Ans. by NTA (3)

(1) B, C and D only

Sol. $\vec{F} = q(\vec{v} \times \vec{B})$ as angle between \vec{v} and \vec{B} is 0°

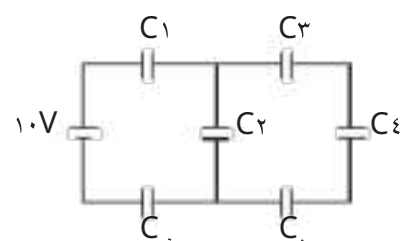
$F = 0$.

SECTION-B

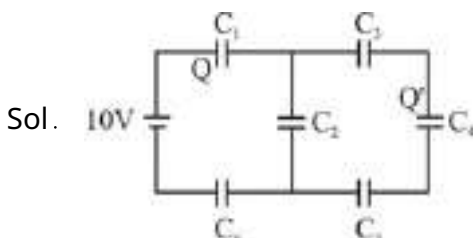
51. In the given circuit,

$C_1 = 2 \mu F$, $C_2 = 0.2 \mu F$, $C_3 = 2 \mu F$, $C_4 = 2 \mu F$,

$C_5 = 2 \mu F$, $C_6 = 2 \mu F$, the charge stored on capacitor C_4 is _____ μC .



Official Ans. by NTA (2)



Sol.

$$C_{eq} = 0.5 \mu F$$

$$Q = 0.5 \times 10 = 5 \mu C$$

$$Q' = \frac{C \times 0.5}{0.5 + 0.5} = 0.25 \mu C$$

52. A circular plate is rotating in horizontal plane, about an axis passing through its center and perpendicular to the plate, with an angular velocity ω . A person sits at the center having two dumbbells in his hands. When he stretches out his hands, the moment of inertia of the system becomes triple. If E be the initial Kinetic energy of the system, then final Kinetic energy will be $\frac{E}{x}$.

The value of x is

Official Ans. by NTA (3)

$$\text{Sol. } KE = \frac{L^2}{2I} \Rightarrow \frac{KE_{\text{final}}}{KE_{\text{initial}}} = \frac{I_{\text{initial}}}{I_{\text{final}}} \Rightarrow \frac{KE_{\text{final}}}{E} = \frac{1}{3}$$

53. A nucleus disintegrates into two nuclear parts, in such a way that ratio of their nuclear sizes is $4:1$. Their respective speed have a ratio of $n:1$. The value of n is _____

Official Ans. by NTA (2)

$$\text{Sol. } \frac{v_1}{v_2} = \frac{m_2}{m_1} = \frac{A_2}{A_1} = \frac{1}{4}$$

54. Two identical cells each of emf $1.5 V$ are connected in series across a 10Ω resistance. An ideal voltmeter connected across 10Ω resistance reads $1.5 V$. The internal resistance of each cell is _____ Ω .

Official Ans. by NTA (6)

$$\text{Sol. } V = I \times 10$$

$$1.5 = \frac{2}{10 + 2r} \times 10$$

$$r = 0 \Omega$$

55. A block of mass 5 kg starting from rest pulled up on a smooth incline plane making an angle of 30° with horizontal with an effective acceleration of 10 m/s^2 delivered by the pulling force at $t = 10 \text{ s}$ from the start is _____ W.

Use $g = 10 \text{ m/s}^2$

(calculate the nearest integer value)

Official Ans. by NTA (300)

Sol. $F = mg \sin 30^\circ = 50 \text{ N}$ $F = 50 + 50 = 100 \text{ N}$

$V_{10} = u + at \Rightarrow V_{10} = 0 + 1(10) = 10 \text{ m/s}$

$P_{10} = Fv = 1000 \text{ W}$

56. A coil has an inductance of 2 H and resistance of $\frac{1}{2} \Omega$. A 10 V is applied across the coil. The energy stored in the magnetic field after the current has built up to its equilibrium value will be _____ J.

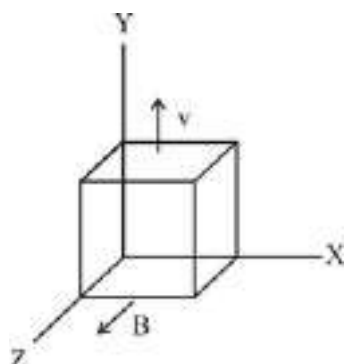
Official Ans. by NTA (250)

Sol. $I = \frac{V}{R} = \frac{10}{\frac{1}{2}} = 20 \text{ A}$

$E = \frac{1}{2} LI^2 = \frac{1}{2} \times 2 \times (20)^2 = 400 \text{ J}$

$E = 250 \text{ J}$

57. A metallic cube of side 10 cm moving along y-axis at a uniform velocity of 2 m/s. In a region of uniform magnetic field of magnitude 0.5 T directed along z-axis. In equilibrium the potential difference between the faces of higher and lower potential developed because of the motion through the field will be _____ mV.



Official Ans. by NTA (150)

Sol. $\mathcal{E} = (v \times B) \cdot d$

$\mathcal{E} = (2 \times 0.5) \times 0.10$

$\mathcal{E} = 0.10 \text{ V}$

58. A wire of density $8 \times 10^{-3} \text{ kg/m}$ is stretched between two clamps 0.8 m apart. The extension developed in the wire is $2.2 \times 10^{-2} \text{ m}$. If $Y = 8 \times 10^{11} \text{ N/m}^2$, the fundamental frequency of vibration in the wire will be _____ Hz.

Official Ans. by NTA (80)

Sol. $f = \frac{1}{2L} \sqrt{\frac{T}{\mu}} = \frac{1}{2L} \sqrt{\frac{YA \Delta L}{AL}}$

$f = 80 \text{ Hz}$

59. The surface tension of soap solution is $2.5 \times 10^{-2} \text{ N/m}$. The amount of work done required to increase the radius of soap bubble from 1 cm to 2 cm is _____ J.

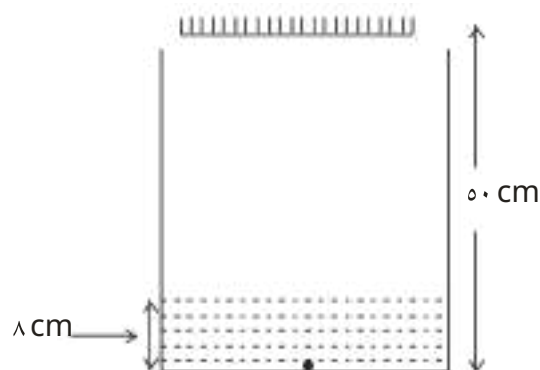
Official Ans. by NTA (264)

Sol. $W = T \cdot (\Delta A)$

$W = T \cdot 8\pi r_1 r_2$

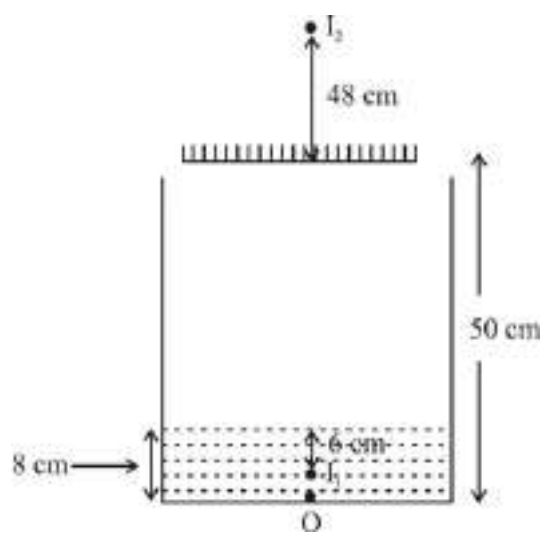
$W = 264 \times 10^{-4} \text{ J}$

60. As shown in the figure, a plane mirror is fixed at a height of 50 cm from the bottom of tank containing water $\frac{4}{3}$. The height of water in the tank is 80 cm. A small bulb is placed at the bottom of the water tank. The distance of image of the bulb formed by mirror from the bottom of the tank is _____ cm.



Official Ans. by NTA (90)

Sol.



Apparent depth of O is

Distance between O and $I_2 = 8 + 48 = 56 \text{ cm}$

CHEMISTRY

SECTION-A

११. Which hydride among the following is less stable:

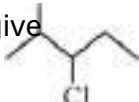
- (१) BeH_2 (२) NH_3
(३) HF (४) LiH

Official Ans. by NTA (१)

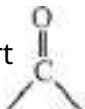
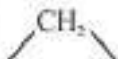
Solution : BeH_2 is hypovalent

१२. Given below are two statements, one is labelled as Assertion A and the other is labelled as Reason R.

Assertion A :  can be subjected to

Wolff-Kishner reduction to give .

Reason R : Wolff-Kishner reduction is used to

convert  into .

In the light of the above statements, choose the correct answer from the options given below :

- (१) Both A and R are true but R is NOT the correct explanation of A.
(२) A is true but R is false.
(३) A is false but R is true.
(४) Both A and R are true and R is the correct explanation of A.

Official Ans. by NTA (३)

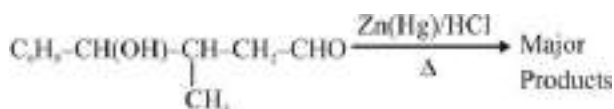
Solution :

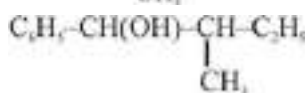
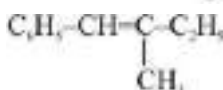
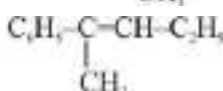
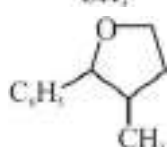
Wolff-Kishner reduction is not suitable for base sensitive group.



TEST PAPER WITH SOLUTION

The major product formed in the following reaction is:



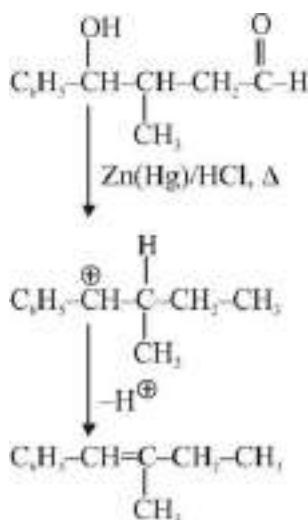
- (A) 
(B) 
(C) 
(D) 

Choose the correct answer from the options given below :

- (१) A only
(२) B only
(३) C only
(४) D only

Official Ans. by NTA (२)

Solution :



٦٤. Which of the following compounds is an example of Freons

- (١) $\text{C}_2\text{Cl}_2\text{F}_2$
 (٢) C_2HF_2
 (٣) $\text{C}_2\text{H}_2\text{F}_2$
 (٤) C_2F_4

Official Ans. by NTA (١)

Solution : Freons are chlorofluoro carbon.

٦٥. For a chemical reaction $\text{A} + \text{B} \rightarrow \text{Product}$, the order is ١ with respect to A and B.

Rate $\text{mol L}^{-1} \text{ s}^{-1}$	A mol L^{-1}	B mol L^{-1}
٠.١٠	٢.٠	٠.٥
٠.٤٠	X	٠.٥
٠.٨٠	٤.٠	Y

What is the value of x and y?

- (١) ٨.٠ and ٢
 (٢) ٤.٠ and ٤
 (٣) ١٦.٠ and ٤
 (٤) ٨.٠ and ٤

Official Ans. by NTA (١)

Solution :

$$r = K[\text{A}]^x[\text{B}]^y$$

$$٠.١ = K(٢.٠)^x(٠.٥)^y \quad \dots (i)$$

$$٠.٤ = K(X)^x(٠.٥)^y \quad \dots (ii)$$

$$٠.٨ = K(٤.٠)^x(Y)^y \quad \dots (iii)$$

From (i) and (ii)

$$X = ٨.٠$$

From (i) and (iii)

$$Y = ٢$$

Given below are two statements. one is labelled as Assertion A and the other is labelled as Reason R.

Assertion A : $[\text{CoCl}(\text{NH}_3)_5\text{H}_2\text{O}]^{2+}$ absorbs at lower wavelength of light with respect to $[\text{Co}(\text{NH}_3)_5(\text{H}_2\text{O})]^{2+}$

Reason R : It is because the wavelength of the light absorbed depends on the oxidation state of the metal ion.

In the light of the above statements, choose the correct answer from the options given below :

- (١) A is false but R is true.
 (٢) A is true but R is false.
 (٣) Both A and R are true and R is the correct explanation of A.
 (٤) Both A and R are true and R is NOT the correct explanation of A.

Official Ans. by NTA (١)

Solution : Since H_2O is strong field ligand compared to chloride and Co^{+2} ion is present.

Δ CFSE is higher for $[\text{Co}(\text{NH}_3)_5\text{H}_2\text{O}]^{2+}$, hence it will absorb at lower wavelength.

٦٧. Given below are two statements. one is labelled as Assertion A and the other is labelled as Reason R.

Assertion A : A solution of the product obtained by heating a mole of glycine with a mole of chlorine in presence of red phosphorous generate chiral carbon atom.

Reason R : A molecule with ٢ chiral carbons is always optically active.

In the light of the above statements, choose the correct answer from the options given below :

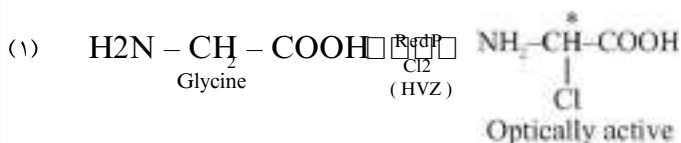
- (١) A is false but R is true.
 (٢) A is true but R is false.
 (٣) Both A and R are true and R is the correct

explanation of A.

- (٤) Both A and R are true and R is NOT the correct explanation of A.

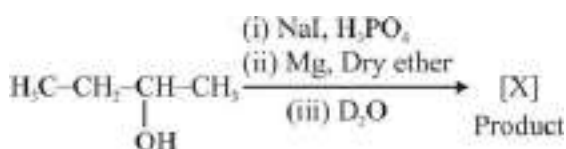
Official Ans. by NTA (٢)

Solution :

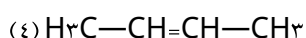
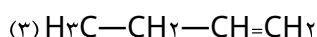
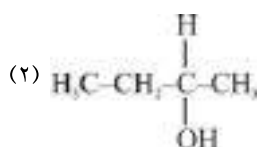
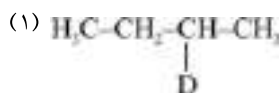


- (٢) Meso compound are optically inactive.

78.

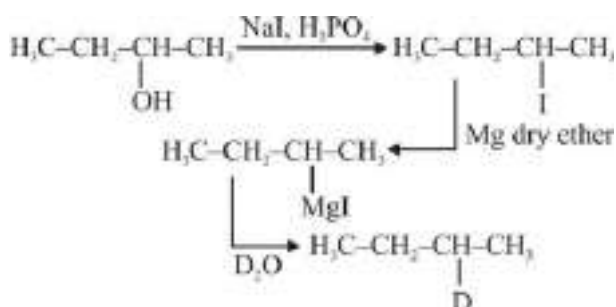


Product [X] formed in the above reaction is:



Official Ans. by NTA (1)

Solution :



79.

Given below are two statements :

Statement I : Ethene at 333 to 343 K and 1-7 atm pressure in the presence of AlEt_3 and TiCl_4 undergoes addition polymerization to give LDP.

Statement II : Caprolactam at 533-543 K in H_2O through step growth polymerizes to give Nylon 6.

In the light of the above Statements, choose the correct answer from the options given below :

(1) Both Statement I and Statement II are true.

(2) Statement I is false but Statement II is true.

(3) Statement I is true but Statement II is false.

(4) Both Statement I and Statement II are false.

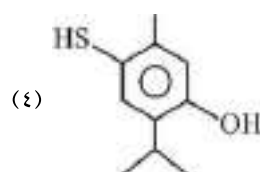
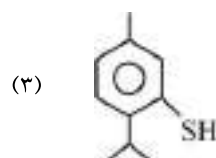
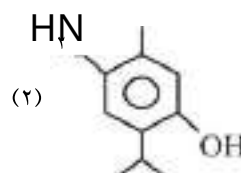
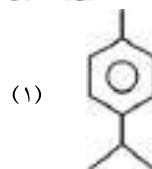
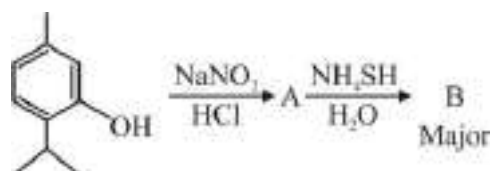
Official Ans. by NTA (2)

Solution : S1 HDPE is formed by TiCl_4 &

AlEt_3 .

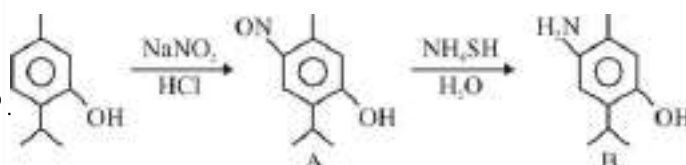
S2 Nylon-6 is formed by caprolactam.

80. Compound B' is



Official Ans. by NTA (2)

Solution :



Which one of the following pairs is an example of polar molecular solids:

(1) $\text{SO}_2(\text{s})$, $\text{NH}_3(\text{s})$

(2) $\text{SO}_2(\text{s})$, $\text{CO}_2(\text{s})$

(3) $\text{HCl}(\text{s})$, $\text{AlN}(\text{s})$

(4) $\text{MgO}(\text{s})$, $\text{SO}_2(\text{s})$

Official Ans. by NTA (1)

Solution : SO_2 and NH_3 are polar molecules. They are constituent particles of polar molecular solids.

72. One mole of P_4 reacts with λ moles of $SOCl_2$ to give ϵ moles of A , x mole of SO_2 and y moles of B . A , B and x respectively are

- (1) PCl_3 , S_2Cl_2 and ϵ
- (2) $POCl_3$, S_2Cl_2 and ϵ
- (3) PCl_3 , S_2Cl_2 and y
- (4) $POCl_3$, S_2Cl_2 and y

Official Ans. by NTA (1)

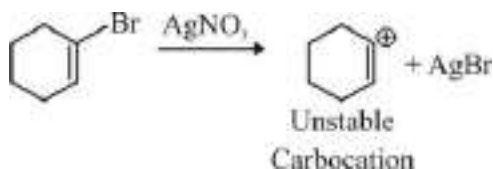
Solution : $P_4 + \lambda SOCl_2 \rightarrow \epsilon PCl_3 + y S_2Cl_2 + \epsilon SO_2$

73. Compound from the following that will not produce precipitate on reaction with $AgNO_3$ is :

- (1)
- (2)
- (3)
- (4)

Official Ans. by NTA (1)

Solution :



74. A solution is prepared by adding y g of $-X||$ of λ mole of water. Mass percent of $-X||$ in the solution is :

- (1) 20%
- (2) 5%
- (3) 2%
- (4) 10%

Official Ans. by NTA (4)

Solution : Solute (X) = y g

Solvent (H_2O) = λ mole = 18λ g

Total mass = $y + 18\lambda = 20$ g

% mass of $X = \frac{y}{20} \times 100 = 10\%$

Given below are two statements : Statement-I : In the metallurgy process, sulphide ore is converted to oxide before reduction. Statement-II : Oxide ores in general are easier to reduce. In the light of the above statements, choose the most appropriate answer from the options given below :

- (1) Both Statement I and Statement II are correct.
- (2) Statement I is correct but Statement II is incorrect.
- (3) Statement I is incorrect but Statement II is correct.
- (4) Both Statement I and Statement II are incorrect.

Official Ans. by NTA (1)

Solution : $yZnS + rO_2 \rightarrow yZnO + rSO_2$

Oxides on carbon reduction forms CO_2 while sulphide on carbon reduction gives CS_2 .

CO_2 is more volatile compared to CS_2 therefore oxides are easy to reduce.

Alkali metal from the following with least melting point is :

- (1) Rb
- (2) K
- (3) Na
- (4) Cs

Official Ans. by NTA (4)

Solution : On moving down the group in alkali metals melting point decreases.

vii. What weight of glucose must be dissolved in 100 g of water to lower the vapour pressure by 1.2 mm Hg? (Assume dilute solution is being formed) Given : Vapour pressure of pure water is 0.2 mm Hg at room temperature. Molar mass of glucose is

180 g mol⁻¹

(1) 4.69 g

(2) 3.09 g

(3) 2.09 g

(4) 3.69 g

Official Ans. by NTA (4)

Solution : $\frac{P_0 - P_s}{P_0} = \frac{n}{N}$ (for dilute solution)

$$\frac{0.2}{0.2} = \frac{n}{100}$$

100

$$\frac{271 \times 18}{100 \times 180}$$

$$w = \frac{100 \times 180}{271 \times 18} ; w = 3.69 \text{ g}$$

viii. The magnetic moment is measured in Bohr Magneton (BM).

Spin only magnetic moment of Fe in $\text{Fe}(\text{H}_2\text{O})_6^{2+}$

and $\text{Fe}(\text{CN})_6^{4-}$ complexes respectively is :

(1) 6.92 B.M. in both

(2) 4.89 B.M. and 6.92 B.M.

(3) 3.87 B.M. and 1.732 B.M.

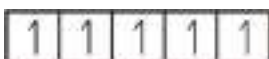
(4) 0.92 B.M. and 1.732 B.M.

Official Ans. by NTA (4)

Solution : $\text{Fe}(\text{H}_2\text{O})_6^{2+}$

Fe: $[\text{Ar}] 3d^6 4s^2$

No pairing



Unpaired e = 0

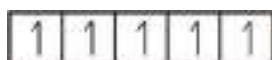
$$\mu = \sqrt{n(n+2)}$$

$$\mu = \sqrt{0(0+2)}$$

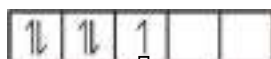
$\mu = 0.92 \text{ B.M.}$

$\text{Fe}(\text{CN})_6^{4-}$

Fe: $[\text{Ar}] 3d^6 4s^2$



Pairing occurs due to strong field ligand CN



Unpaired e = 1

$$\mu = \sqrt{n(n+2)}$$

$\mu = 1.732 \text{ B.M.}$

viii. Match List I with List II.

List I		List II	
Complex		Colour	
A.	$\text{Mg}(\text{NH}_4)\text{PO}_4$	I.	Brown
B.	$\text{K}_2\text{Co}(\text{NO}_2)_6$	II.	White
C.	$\text{MnO}(\text{OH})_2$	III.	Yellow
D.	$\text{Fe}_3\text{Fe}(\text{CN})_6$	IV.	blue

Choose the correct answer from the options given below : (1) A-II, B-III, C-I, D-IV (2) A-III, B-IV, C-II, D-I (3) A-II, B-IV, C-I, D-III (4) A-II, B-III, C-IV, D-I Official Ans. by NTA (1)

Solution : $\text{Mg}(\text{NH}_4)\text{PO}_4$: White

$\text{K}_2\text{Co}(\text{NO}_2)_6$: Yellow

$\text{MnO}(\text{OH})_2$: Brown

$\text{Fe}_3\text{Fe}(\text{CN})_6$: Blue

ix.

If Ni is replaced by Pt in the complex

NiCl_2Br_2 , which of the following properties are expected to get changed?

A. Geometry

B. Geometrical isomerism

C. Optical isomerism

D. Magnetic properties

(1) A, B and C

(2) A, B and D

(3) A and D

(4) B and C

Official Ans. by NTA (2)

Solution : NiBr_2Cl_2 : This complex species is tetrahedral as Br and Cl are weak field ligands.

PtBr_2Cl_2 : As Pt belongs to d series. This complex species is square planar.

Both the complex species are optically inactive.

NiBr_2Cl_2 , being tetrahedral does not show Geometrical Isomerism.

PtBr_2Cl_2 shows two Geometrical Isomers.

SECTION-B

81. Number of compounds from the following which will not produce orange red precipitate with

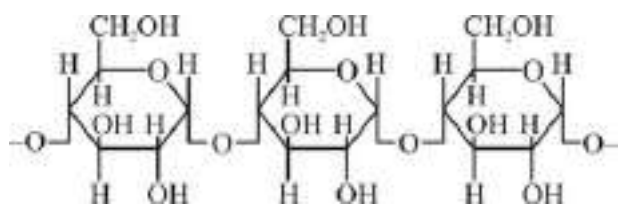
Benedict solution is

Glucose, maltose, sucrose, ribose, γ -deoxyribose, amylose, lactose.

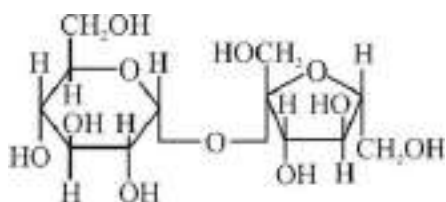
Official Ans. by NTA (r)

Solution :

Amylose



Sucrose :



Both Amylose and Sucrose does not give Benedict's test. \therefore moles each of hydrogen and iodine is heated in

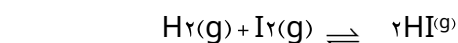
82. a sealed ten litre vessel. At equilibrium, γ

~~moles of~~

~~found. The equilibrium constant for~~

Official Ans. by NTA (v)

Solution :



t = 0 1.0 1.0 -

t_{eq} x x 2

$$K_c = \frac{[\text{HI}]^2}{[\text{H}_2][\text{I}_2]} = \frac{(2)^2}{(1-x)(1-x)} = 4$$

83.

The number of correct statements about modern adsorption theory of heterogeneous catalysis from the following is

A. The catalyst is diffused over the surface of reactants.

B. Reactants are adsorbed on the surface of the catalyst.

C. Occurrence of chemical reaction on the catalyst's surface through formation of an intermediate.

D. It is a combination of intermediate compound formation theory and the old adsorption theory.

E. It explains the action of the catalyst as well as those of catalytic promoters and poisons.

Official Ans. by NTA (r)

Solution : B, C and D are correct.

(NCERT – Surface Chemistry)

84.

The number of correct statements from the following _____

A. For $1s$ orbital, the probability density is maximum at the nucleus.

B. For $1s$ orbital, the probability density first increases to maximum and then decreases sharply to zero.

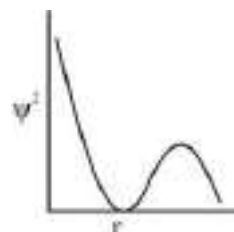
C. Boundary surface diagrams of the orbitals encloses a region of 100% probability of finding the electron.

D. p and d-orbitals have 1 and 2 angular nodes respectively.

E. Probability density of p-orbital is zero at the nucleus.

Official Ans. by NTA (r)

Solution : A, D and E statements are correct.



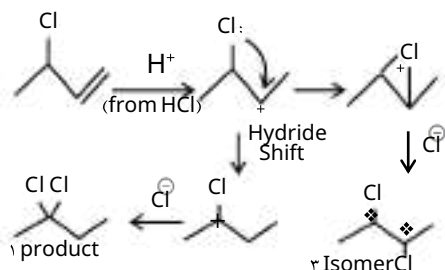
For $1s$ orbital, the probability density first decreases and then increases.

~~At nucleus the probability~~

density of finding electron is never zero and it always have some finite value.

86. The number of possible isomeric products formed when 3-chloro-1-butene reacts with HCl through carbocation formation is _____ Official Ans. by NTA (3)

Solution :



Total Possible Isomeric product = 1 + 2 = 3

87. $\text{Mg}(\text{NO}_3)_2 \cdot x\text{H}_2\text{O}$ and $\text{Ba}(\text{NO}_3)_2 \cdot y\text{H}_2\text{O}$ represent formula of the crystalline forms of nitrate salts. Sum of X and Y is _____ Official Ans. by NTA (6)

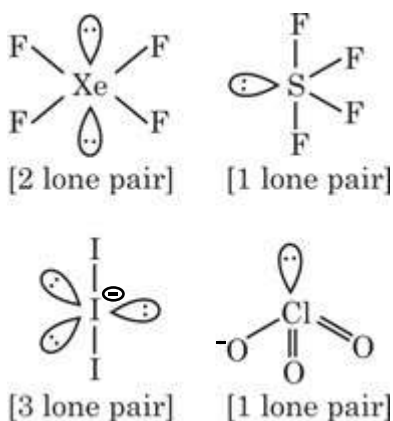
Solution : $\text{Mg}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ is a hydrated salt whereas $\text{Ba}(\text{NO}_3)_2$ is an anhydrous salt.
 $x + y = 6$

88. The total number of intensive properties from the following is _____
 Volume, Molar heat capacity, Molarity, E_{cell} , Gibbs free energy change, Molar mass, Mole
 Official Ans. by NTA (3)

Solution : Extensive \square Mole, Volume, Gibbs free energy.
 Intensive \square Molar mass, Molar heat capacity, Molarity, E_{cell} .

89. The maximum number of lone pairs of electrons on the central atom from the following species is _____
 ClO_2^- , XeF_4 , SF_6 and IF_7
 Official Ans. by NTA (3)

Solution :



90. The volume of hydrogen liberated at STP by treating 2.4 g of magnesium with excess of hydrochloric acid is _____ L.
 Given: Molar volume of gas is 22.4 L at STP.

Molar mass of magnesium is 24 g mol.
 Official Ans. by NTA (22.4)

Solution :



$$w = 2.4 \text{ g}$$

$$N = \frac{2.4}{24} = 0.1 \text{ mole}$$

$$1 \text{ mole of gas at STP} \square 22.4 \text{ lit.}$$

$$\square 0.1 \text{ mole of gas} = 0.1 \times 22.4$$

$$= 2.24 \text{ lit.} = 22.4 \times 10^{-2} \text{ litre}$$

91. The number of correct statements from the following is :

A. Ecell is an intensive parameter.

B. A negative E° means that the redox couple is a stronger reducing agent than the H^+/H_2 couple.

C. The amount of electricity required for oxidation or reduction depends on the stoichiometry of the electrode reaction.

D. The amount of chemical reaction which occurs at any electrode during electrolysis by a current is proportional to the quantity of electricity passed through the electrolyte.

Official Ans. by NTA (3)

Solution : Given statements A, B, C and D are correct.